

全英语课程采用如下英文版教学大纲

## Syllabus

**Department:** School of Life Sciences

**Date:** February 17, 2024

<b>Course Code</b>	BIOL130093.01						
<b>Course Title</b>	Biological Thermodynamics						
<b>Credit</b>	2	<b>Experiment (including Computer) Credit</b>	0	<b>Practice Credit</b>	0	<b>Aesthetic Education Credit</b>	0
<b>Credit Hours Per Week</b>	2	<b>Education on The Hard-Working Spirit Credit Hours</b>	0	<b>Language of Instruction</b>	English	<b>Honors Course</b>	<input type="checkbox"/> Yes <input checked="" type="checkbox"/> No
<b>Course Nature</b>	<input type="checkbox"/> Core General Education Course <input type="checkbox"/> Specific General Education Course <input type="checkbox"/> Basic Course in General Discipline <input type="checkbox"/> Others			2+X Major : <input type="checkbox"/> Professional Core Course <input checked="" type="checkbox"/> Professional Advanced Course Non 2+X Major : <input type="checkbox"/> Professional Compulsory Course <input type="checkbox"/> Professional Elective Course			
<b>Course Objectives</b>	Thermodynamic principles are widely applied in biological research work, especially in studying the structures and functions of biomolecules. The aim of this course is to help students understand the basic thermodynamic principles and apply these principles in biological research work, e.g., in the field of drug design.						
<b>Course Description</b>	Course lectures include the introductions of the basic concepts of thermodynamics and their applications in biology. These concepts include the First Law and the Second Law of Thermodynamics, and the related principles including chemical equilibria, phase equilibria, binding equilibria, etc. In addition, the basic concepts of kinetics will be introduced along with their applications in biology. Towards the end of the course, there is a brief introduction of computational drug design with the examples utilizing the principles of thermodynamics and kinetics. The course is designed to help students understand basic principles of thermodynamics and kinetics and apply them in biological research work.						
<b>Course Requirements:</b> None.							

**Teaching Methods:** All lectures will be given in English.

**Course Director's Academic Background:**

The instructor works in the fields of biophysics, computational biology, and thermodynamics.

**Instructor's Academic Background:** Same as above.

**Members of Teaching Team**

Name	Gender	Professional Title	Department	Responsibility
Di Wu	Female	Associate Researcher	School of Life Sciences	Instructor

**Course Schedule** (Please supply the details about each lesson):

(Please note that some of the following lectures may be joined together in case of the holiday break.)

Lecture 1: Introduction

Lecture 2: Introduction of the Laws of Thermodynamics Using Biological Examples

Lecture 3: Heat, Work, Energy, and Biofuels

Lecture 4: Enthalpy, Heat Capacity, and Their applications in Biology

Lecture 5: Entropy, and Its Applications in Biology

Lecture 6: The Second Law of Thermodynamics, and Its Applications in Biology

Lecture 7: Free Energy, and Its Applications in Biology

Lecture 8: Chemical Equilibria

Lecture 9: Phase Equilibria

Lecture 10: Solutions and Mixtures

Lecture 11: Stability of Protein Structures

Lecture 12: Introduction to Kinetics

Lecture 13: Applications of Kinetics in Biology

Lecture 14: Ligand Binding

Lecture 15: Thermodynamics and Kinetics of Drug Binding and a Brief Introduction to Computational Drug Design

Lecture 16: Presentations by the Students

**The design of class discussion or exercise, practice, experience and so on:** NA

**If you need a TA, please indicate the assignment of assistant:** NA

**Grading & Evaluation** (Provide a final grade that reflects the formative evaluation process):

- Homework: 20%
- Attendance and Participation: 20%
- Project Presentation: 25%
- Project Paper: 35%

**Usage of Textbook:** ☐ Yes (complete textbook information form below) ☒ No

**Textbook Information** (No more than two textbooks) :

Title	Author	ISBN	Publishing time	Publisher	Type <input type="checkbox"/>	Type <input type="checkbox"/>
					<input type="checkbox"/> Self-compiled Textbook (Published) <input type="checkbox"/> Non-mainland Textbook <input type="checkbox"/> Other Textbook (Published)	<input type="checkbox"/> National Planning Textbook <input type="checkbox"/> Provincial and Ministerial Planning Textbook <input type="checkbox"/> School Level Planning Textbook <input type="checkbox"/> Others
					<input type="checkbox"/> Self-compiled Textbook (Published) <input type="checkbox"/> Non-mainland Textbook <input type="checkbox"/> Other Textbook (Published)	<input type="checkbox"/> National Planning Textbook <input type="checkbox"/> Provincial and Ministerial Planning Textbook <input type="checkbox"/> School Level Planning Textbook <input type="checkbox"/> Others

**Teaching References** (Including author, title, publisher, publishing time, ISBN):

- Donald T. Haynie, Biological Thermodynamics, Cambridge University Press, 2001. (ISBN: 9780521884464).
- Ken A. Dill & Sarina Bromberg, Molecular Driving Forces: Statistical Thermodynamics in Chemistry and Biology, Garland Science, 2003. (ISBN: 9780815320517).
- Gordon G. Hammes, Thermodynamics and Kinetics for the Biological Sciences, John Wiley & Sons, 2000. (ISBN: 0471374911).

Table column size can be adjusted according to the content.